

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 48 (2007) 5539-5541

## Synthesis of monofluorinated 1-(naphthalen-1-yl)piperazines

Joseph T. Repine,<sup>a,\*</sup> Douglas S. Johnson,<sup>a,\*</sup> Andrew D. White,<sup>a</sup> David A. Favor,<sup>a</sup> Michael A. Stier,<sup>a</sup> Judy Yip,<sup>b</sup> Trent Rankin,<sup>b</sup> Qizhu Ding<sup>b</sup> and Samarendra N. Maiti<sup>b</sup>

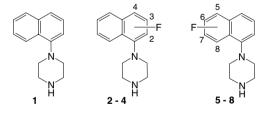
<sup>a</sup>Pfizer Global Research and Development, 2800 Plymouth Road, Ann Arbor, MI 48105, USA <sup>b</sup>Naeja Pharmaceuticals, 4290-91a Street, Edmonton, Alberta, Canada T6E 5V2

> Received 4 April 2007; revised 24 May 2007; accepted 25 May 2007 Available online 2 June 2007

Abstract—A series of regioisomerically monofluorinated 1-(naphthalen-1-yl)piperazines is described. © 2007 Elsevier Ltd. All rights reserved.

4-Arylpiperazines are considered 'privileged' templates in drug discovery.1 This core structure is found in numerous biologically active compounds and drugs, which act by various mechanisms for diverse indications. Foremost among these is their activity against monoamine based GPCRs, namely, dopaminergic,<sup>2</sup> serotonergic,<sup>3</sup> and adrenergic<sup>4</sup> GPCRs. Subtle structural changes of the arylpiperazine moiety in a lead molecule are often desired in order to optimize the potency and selectivity against these receptors and modulate the physical and ADME (absorption, distribution, metabolism, and elimination) properties of the molecule. The incorporation of fluorine into a lead molecule can alter its chemical properties (Pauling electronegativity of fluorine is 3.98 vs 2.20 for hydrogen), pharmacokinetic and pharmacodynamic properties (metabolic stability and absorption), and biological activity.<sup>5a</sup> It has been recognized that fluorine substitution on aromatic rings can be used to redirect metabolism while often preserving or improving receptor binding of medicinal compounds.<sup>5b</sup> Furthermore, aromatic fluorine atoms can participate in hydrogen bonding interactions.<sup>5c</sup> Therefore, fluoroaromatic piperazines are highly desirable intermediates for the synthesis of new drug candidates.

We describe herein the synthesis of a series of monofluorinated 1-(naphthalen-1-yl)piperazines (1), including 2-fluoronaphthalen-1-yl (2), 3-fluoronaphthalen-1-yl (3), 4-fluoronaphthalen-1-yl (4), 5-fluoronaphthalen-1-yl (5), 7-fluoronaphthalen-1-yl (7), and 8-fluoronaphthalen-1-yl piperazines (8).<sup>6a,b</sup> While the 6- and 7-fluoronaphthalen-1-yl piperazines (**6** and **7**) have been disclosed previously,<sup>7a-c</sup> we report the details of a more regioselective synthesis of intermediates leading to **7** (Scheme 5). The known methods for the preparation of arylpiperazines include reacting arylamines with bis(2-chloro-2-ethyl)amine<sup>8a</sup> or amination of aryl halides or triflates with monoprotected piperazine under Buchwald–Hartwig–Migita cross-coupling conditions.<sup>8b-d</sup> Selection of these techniques in combination with preparation of various functionalized fluoronaphthyl intermediates provided routes to the complete set of monofluorinated 1-(naphthalen-1-yl)piperazines **2–8**.

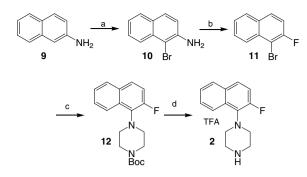


The synthesis of 1-(2-fluoronaphthalen-1-yl)piperazine<sup>6a,b</sup> (2) began with selective bromination of naphthalen-2-amine (9) at the 1-position to give 10 (Scheme 1).<sup>9</sup> Diazotization of amine 10 in the presence of fluoroboric acid provided the intermediate diazonium salt, which was isolated and subjected to thermal decomposition to yield the desired 1-bromo-2-fluoronaphthalene (11). Amination of the bromide 11 with Boc-piperazine followed by deprotection of the Boc group with trifluoroacetic acid provided 2 as the TFA salt.

In a manner similar to that described by Adcock and Dewar,<sup>10</sup> the synthesis of 1-(3-fluoronaphthalen-1-yl)piperazine (3) commenced with the diazotization of

<sup>\*</sup> Corresponding authors. E-mail addresses: aarepinej@comcast.net; doug.johnson@pfizer.com

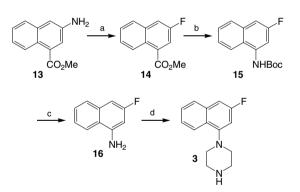
<sup>0040-4039/\$ -</sup> see front matter @ 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2007.05.156



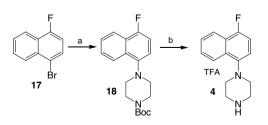
Scheme 1. Reagents and conditions: (a) NBS, DMF, 110 °C, 70%; (b) HBF<sub>4</sub>, NaNO<sub>2</sub>, THF, NaBF<sub>4</sub>; xylenes, reflux, 1 h, 47%; (c) Pd(OAc)<sub>2</sub>, NaO'Bu, 2-(dicyclohexylphosphino)biphenyl, Boc-piperazine, toluene, 80 °C, 40%; (d) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 40%.

methyl 3-amino-1-naphthoate (13) in the presence of fluoroboric acid to give methyl 3-fluoro-1-naphthoate (14) (Scheme 2). Hydrolysis of the ester followed by Curtius rearrangement of the acid and trapping of the resultant isocyanate with *t*-butanol afforded the Boc-protected naphthylamine 15. Removal of the Boc group with trifluoroacetic acid provided 3-fluoronaphthalen-1-amine (16). Finally, the piperazine was constructed by reacting amine 16 with bis(2-chloro-2-ethyl)amine to give 3.

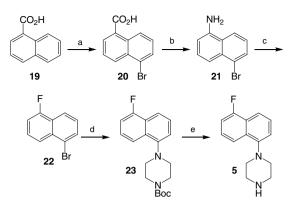
A facile route to 1-(4-fluoronaphthalen-1-yl)piperazine (4) is shown in Scheme 3. *N*-Arylation of Boc-piperazine with commercially available 1-bromo-4-fluoronaphthalene (17, Aldrich) yielded 18 and subsequent Boc cleav-



Scheme 2. Reagents and conditions: (a) 48% HBF<sub>4</sub>, NaNO<sub>2</sub>, THF, NaBF<sub>4</sub>; PhCl, reflux, 1 h, 50%; (b) (1) 2 N KOH, MeOH, reflux, 2 h, 98%; (2) DPPA, Et<sub>3</sub>N, 'BuOH, 87%; (c) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 33%; (d) bis-(2-chloro-2-ethyl)amine hydrochloride, <sup>i</sup>Pr<sub>2</sub>EtN, PhCl, 150 °C, 12%.



Scheme 3. Reagents and conditions: (a)  $Pd(OAc)_2$ , NaO'Bu, 2-(dicyclohexylphosphino)biphenyl, Boc-piperazine, toluene, 80 °C; (b) TFA/CH<sub>2</sub>Cl<sub>2</sub>, 50:50, 0–25 °C, 1 h, 78% two steps.

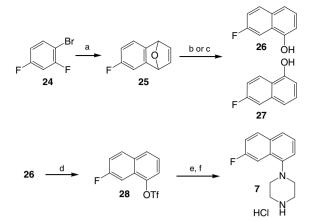


Scheme 4. Reagents and conditions: (a)  $Br_2$ , HOAc, 110 °C, 62%; (b) (1) 'BuOH, Et<sub>3</sub>N, DPPA; 2. TFA, 58%; (c) 48% HBF<sub>4</sub>, NaNO<sub>2</sub>, THF; NaBF<sub>4</sub>, 47%; (d) Pd(OAc)<sub>2</sub>, NaO'Bu, 2-(dicyclohexylphosphino)biphenyl, Boc-piperazine, toluene, 80 °C; (e) 6 N HCl, 66% two steps.

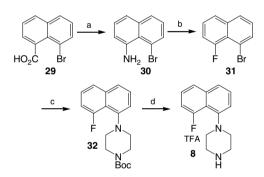
age with 50% TFA in dichloromethane gave 4 in 78% yield over two steps.

The synthesis of 1-(5-fluoronaphthalen-1-yl)piperazine (5) is shown in Scheme 4. Naphthalene-1-carboxylic acid (19) was brominated with a degree of regioselectivity at the 5-position giving 20 as an easily isolable solid.<sup>11</sup> From this acid, Tagat et al. formed the *t*-butyl ester and converted the bromine to a fluorine via a halogen-metal exchange followed by quenching with an electrophilic fluorinating agent. Instead, by analogy with chemistry in Scheme 2, we converted carboxylic acid 20 to amine 21, which was subsequently diazotized in the presence of fluoroboric acid to give the fluoro derivative 22. Finally, amination of 22 with Boc-piperazine provided 23, which following deprotection gave 5.

The synthesis of 1-(7-fluoronaphthalen-1-yl)piperazine (7) from 7-fluoronaphthyl triflate (28) (prepared from 7-fluorotetralone) has been disclosed.<sup>7</sup><sup>c</sup> Another route to 7-fluoronaphthalen-1-ol (26) (albeit without experimental detail)<sup>12a</sup> was via acid catalyzed opening of 7-fluoro-1,4-epoxy-1,4-dihydronaphthalene (25). To expedite the synthesis of 7, we desired to improve the ring opening of 25 to the prerequisite 7-fluoronaphthalen-1-ol (26). Compound 25 was obtained from the Diels-Alder cyclization of furan with the benzyne intermediate generated from 1-bromo-2,4-difluorobenzene (24) (Scheme 5) similar to the method of Caster et al.,<sup>12b</sup> except that we used magnesium (THF, reflux) instead of n-BuLi (diethyl ether, -78 °C) to generate the benzyne intermediate. Acid catalyzed openings of the 1,4-epoxy bridge of 1,4-epoxy-1,4-dihydronaphthalene like 25 to a naphthol are known<sup>12c</sup> and substituent directed regioselectivity have been observed.<sup>12d</sup> We desired to find conditions to effect a regioselective opening of the 1,4-epoxy bridge in 25 to give the desired 7-fluoro isomer 26. Initially, we found that treatment of 25 with concentrated HCl/ethanol at reflux gave a 2:1 mixture of the 7- and 6-fluoronaphthalen-1-ols (26 and 27) respectively. However, we discovered that the opening could be directed by use of BF<sub>3</sub>-etherate in dichloromethane giving a 56% yield of the 7-fluoro isomer 26 in 95% regioisomeric purity. Naphthol 26 was converted to triflate 28 and then



Scheme 5. Reagents and conditions: (a) Mg/THF, furan, 2 h, 65 °C, 15 h, 97%; (b) concentrated HCl/ethanol, reflux, 2 h, 68%; (c) BF<sub>3</sub>·OEt<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 1.5 h, 60%; (d) Et<sub>3</sub>N, Tf<sub>2</sub>O CH<sub>2</sub>Cl<sub>2</sub>, rt, 1 h, 99%; (e) 0.1 equiv Pd(OAc)<sub>2</sub>, 1.5 equiv NaO'Bu, 0.12 equiv 2-(dicyclophosphino)biphenyl, 1.5 equiv Boc-piperazine, toluene, 85 °C, 3 h, 80%; (f) HCl (gas), CH<sub>2</sub>Cl<sub>2</sub>, rt, 3 h, 96%.



Scheme 6. Reagents and conditions: (a)  $H_2SO_4$ ,  $NaN_3$ ,  $CHCl_3$ , 40 °C, 96%; (b) 48% HBF<sub>4</sub>,  $NaNO_2$ , THF;  $NaBF_4$ , 69%; (c) Pd(OAc)<sub>2</sub>, NaO'Bu, 2-(dicyclohexylphosphino)biphenyl, Boc-piperazine, toluene, 80 °C, 38%; (d) TFA,  $CH_2Cl_2$ , 0 °C, 89%.

coupled to Boc-piperazine under Buchwald–Hartwig– Migita conditions to give 7 after deprotection of the Boc group.

The synthesis of 1-(8-fluoronaphthalen-1-yl)piperazine (8) (Scheme 6) was analogous to the synthesis of 5 (Scheme 4), except that the starting 1-bromo-8-naphthoic acid (29),<sup>†</sup> was commercially available. Carboxylic acid 29 was converted to amine 30 via a Curtius rearrangement using sodium azide and sulfuric acid. Diazotization followed by fluorination of the diazonium salt with fluoroboric acid provided the requisite 1-bromo-8-fluoronaphthalene (31), which was coupled with Boc-piperazine under Buchwald–Hartwig–Migita conditions to give 32. Finally acidic deprotection with trifluoroacetic acid afforded 1-(8-fluoronaphthalen-1-yl)-piperazine (8).

In summary, a collection of monofluorinated 1-(naphthalen-1-yl)piperazines 2–8 were prepared as templates for drug discovery from a variety of routes and starting materials.

## Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet. 2007.05.156.

## **References and notes**

- 1. Coudert, P.; Rubat, C.; Bastide, M.; Malhuret, R.; Chopineau, J. J. Pharm. Belg. 1995, 50, 445–454.
- (a) Šoskic, V.; Joksimovic, J. Curr. Med. Chem. 1998, 5, 493–512; (b) Zhang, A.; Kan, Y.; Li, F. Exp. Opin. Ther. Pat. 2006, 16, 587–630; (c) Zhang, A.; Neumeyer, J. L.; Baldessarini, R. J. Chem. Rev. 2007, 107, 274–302.
- Pessoa-Mahana, H.; Araya-Maturana, R.; Claudio, S. B.; Pessoa-Mahana, C. D. *Mini-Rev. Med. Chem.* 2003, *3*, 77– 93.
- Manetti, F.; Corelli, F.; Strappaghetti, G.; Botta, M. Curr. Med. Chem. 2002, 9, 1303–1321.
- (a) Ismail, F. M. D. J. Fluorine Chem. 2002, 118, 27–33;
  (b) Park, B. K.; Kitteringham, N. R.; O'Neill, P. M. Ann. Rev. Pharmacol. Toxicol. 2001, 41, 443–470; (c) Razgulin, A. V.; Mecozzi, S. J. Med. Chem. 2006, 49, 7902–7906.
- (a) Clark, J. D.; Davis, J. M.; Favor, D.; Fay, L. K.; Franklin, L.; Henegar, K. E.; Johnson, D. S.; Nichelson, B. J.; Ou, L.; Repine, J. T.; Walters, M. A.; White, A. D.; Zhu, Z. Patent Application US 2005043309, 2005; (b) Johnson, D. S.; Repine, J. T.; White, A. D. US 7,160,888, 2007.
- (a) Frost, J.; Gaudilliere, B.; Rousseau, J.; Dupont, R.; Manoury, P.; Obitz, D.; Eur. Pat. Appl. EP 343050, 1989; (b) Agejas-Chicharro, J.; Bueno Melendo, A. B.; Camp, N. P.; Gilmore, J.; Jimenez-Aguado, A. M.; Lamas-Peteira, C.; Marcos-Llorente, A.; Mazanetz, M. P.; Montero Salgado, C.; Timms, G. H.; Williams, A. C.; PCT Int. Appl. WO 2002050067, 2002; (c) Torrado, A.; Lamas, C.; Agejas, J.; Jimenez, A.; Diaz, N.; Gilmore, J.; Boot, J.; Findlay, J.; Hayhurst, L.; Wallace, L.; Broadmore, R.; Tomlinson, R. *Bioorg. Med. Chem.* 2004, *12*, 5277–5295.
- (a) Glennon, R. A.; Slusher, R. M.; Lyon, R. A.; Titeler, M.; McKenney, J. D. J. Med. Chem. 1986, 29, 2375–2380;
   (b) Wolfe, J. P.; Tomori, H.; Sadighi, J. P.; Yin, J.; Buchwald, S. L. J. Org. Chem. 2000, 65, 1158–1174; (c) Wolfe, J. P.; Wagaw, S.; Marcoux, J.-F.; Buchwald, S. L. Acc. Chem. Res. 1998, 31, 805–818; (d) Hartwig, J. F. Angew. Chem., Int. Ed. 1998, 37, 2046–2067.
- Beugelmans, R.; Chbani, M. Bull. Soc. Chim. Fr. 1995, 132, 290–305.
- Adcock, W.; Dewar, M. J. S. J. Am. Chem. Soc. 1967, 89, 386–390.
- Tagat, J. R.; McCombie, S. W.; Nazareno, D. V.; Boyle, C. D.; Kozlowski, J. A.; Chackalamannil, S.; Josien, H.; Wang, Y.; Zhou, G. J. Org. Chem. 2002, 67, 1171–1177.
- (a) Brun, P.; Guglielmetti, R.; Anguille, S. Appl. Organomet. Chem. 2002, 16, 271–276; (b) Caster, K. C.; Keck, C. G.; Walls, R. D. J. Org. Chem. 2001, 66, 2932–2936; (c) Gribble, G. W.; LeHoullier, C. S.; Sibi, M. P.; Allen, R. W. J. Org. Chem. 1985, 50, 1611–1616; (d) Schlosser, M.; Castagnetti, E. Eur. J. Org. Chem. 2001, 3991–3997.

<sup>&</sup>lt;sup>†</sup> 1-Bromo-8-naphthoic acid, CO-0063, Carbocore Research Chemicals and Intermediates.